sodium acetate was added and the porphin was re-extracted into benzene (3 × 200 ml.). The benzene solution was distilled to dryness under reduced pressure, the residue redissolved in hot benzene (50 ml.) and purified as previously described.

Anal. Calcd. for $C_{20}H_{14}N_4$: C, 77.5; H, 4.5; N, 18.1. Found: C, 77.5; H, 4.7; N, 18.3.

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Substituted Aminobenzacridines

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The preparation of a number of substituted 7-aminobenz[*c*]acridines as potential amoebicides has been reported by Elslager and co-workers[1,2] and by Short and co-workers.[3] The present communication deals with the preparation of a number of substituted 7-aminobenz[*c*]acridines and 12-aminobenz[*a*]acridines for trials against *Entamoeba histolytica in vitro.*

The compounds were prepared by the interaction of 7-chlorobenz[*c*]acridine and 12-chlorobenz[*a*]acridine with a slight excess of the appropriate amine in phenol at 120° for 2 hours and isolated as the salicylate as described in an earlier communication by Chatterjee.[4] The compounds were purified by crystallization from 90% ethanol and are shown in Table I.

TABLE I[a]

Serial No.	Base	Salt	M.p. of Salt, °C.	Carbon, %		Hydrogen, %		Nitrogen, %	
				Calcd	Found	Calcd.	Found	Calcd.	Found
1	7-benzylaminobenz[*c*]acridine	1.5 $C_7H_6O_3$[b]	202	76.52	76.81	4.99	4.98	5.18	5.10
2	12-benzylaminobenz[*a*]acridine	1.5 $C_7H_6O_3$	216	76.52	76.49	4.99	4.80	5.18	5.15
3	7-(2-phenylethyl)aminobenz[*c*]acridine	1.5 $C_7H_6O_3$	204	76.76	76.55	5.23	4.92	5.05	4.90
4	12-(2-phenylethyl)aminobenz[*a*]acridine	1.5 $C_7H_6O_3$	203	76.76	76.91	5.23	4.80	5.05	5.15
5	7-(3-phenylpropyl)aminobenz[*c*] acridine	1.5 $C_7H_6O_3$	187	76.98	77.21	5.45	5.53	4.92	4.75
6	12-(3-phenylpropyl)aminobenz[*a*]acridine	1.5 $C_7H_6O_3$	194	76.98	77.15	5.45	5.55	4.92	4.70
7	7-(4-phenoxybutyl)aminobenz[*c*] acridine	$C_7H_6O_3$	174	76.98	77.00	5.66	5.87	5.28	5.19
8	12-(4-phenoxybutyl)aminobenz[*a*] acridine	$C_7H_6O_3$	172	76.98	76.75	5.66	5.82	5.28	5.20
9	7-*p*-dimethylaminoanilinobenz[*c*]acridine	1.5 $C_7H_6O_3$	200	74.74	74.50	5.26	5.00	7.37	7.26
10	12-*p*-dimethylaminoanilinobenz[*a*] acridine	1.5 $C_7H_6O_3$	196	74.74	74.53	5.26	5.13	7.37	7.42

[a] All melting points are uncorrected. [b] $C_7H_6O_3$, salicylic acid.

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Attempted Preparation of Benzpinacol Carbonate

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In an endeavor to synthetize benzpinacol carbonate, needed for a kinetic study, two different methods have been attempted: 1) a base-catalyzed ester-interchange between benzpinacol (I) and diethyl carbonate, and 2) the reaction of phosgene with I in presence of antipyrine, according to the method of Ludwig and Piech.[1] Both methods failed to produce the desired cyclic carbonate. Instead, the first method gave a mixture consisting of benzophenone (II), ethyl benzhydryl, and dibenzhydryl carbonates (III and IV), whereas the sec-

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